# ALL THAT GLITTERS IS NOT GOLD: CHARACTERIZATION OF METALLIC DECORATIVE ELEMENTS FROM FIVE ENCLOSED GARDENS

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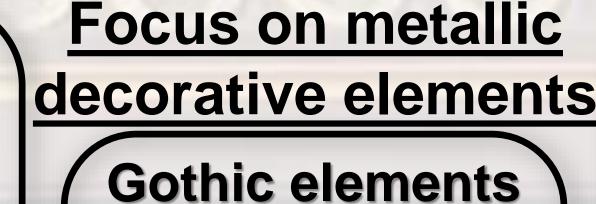
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Gilded wooden sculptures Gilded polychrome

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wooden sculptures Often degraded



Shiny/glossy thin foils Pristine conservation state ❖ Gilded?







panels Well-preserved Similar to back-plates

#### WHY OUR RESEARCH?

☐ For conservation purposes ☐ To acquire more historical and technical information

sculptures

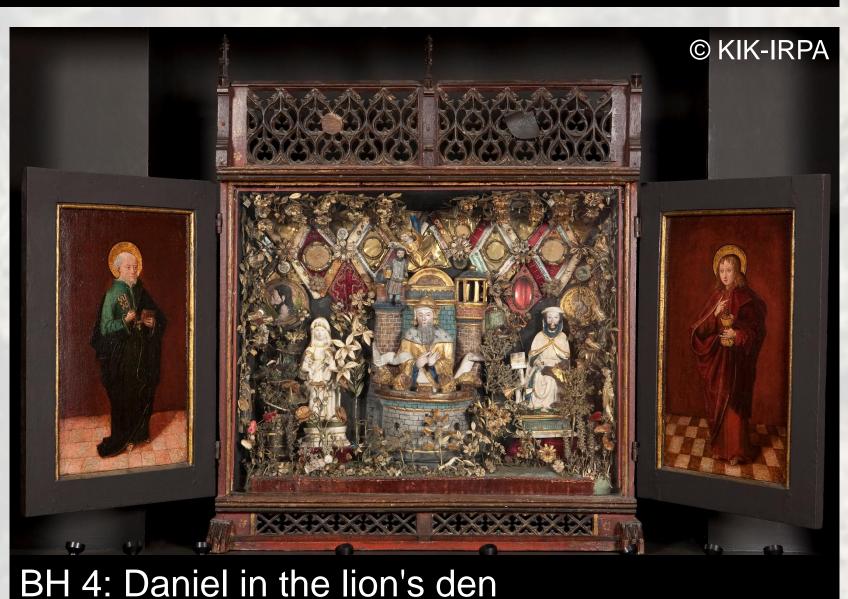


Differences between panels? 2. Are back-plates and sequins

Laboratory analysis on

gilded?





Enameled

1 Cu alloys

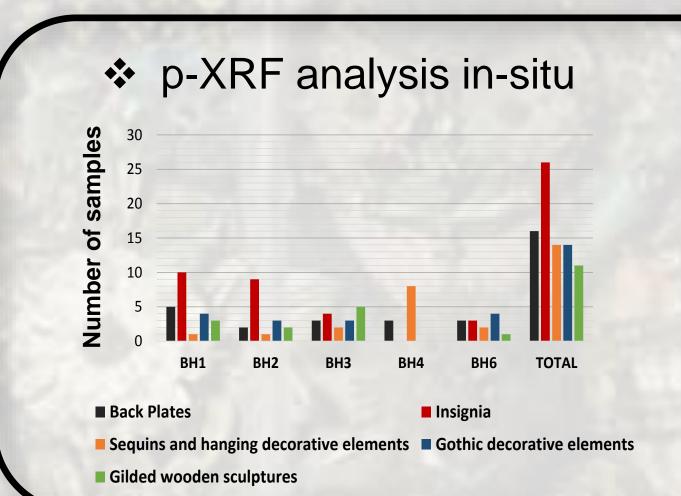


Pb-Sn alloys

☐ 5 reliquary altarpieces ☐ 16th century ☐ Preserved by the sisters of the Onze-Lieve-Vrouwegasthuis of Malines (Belgium)

☐ Huge diversity ☐ Several wellpreserved golden metallic elements





five loose sequins

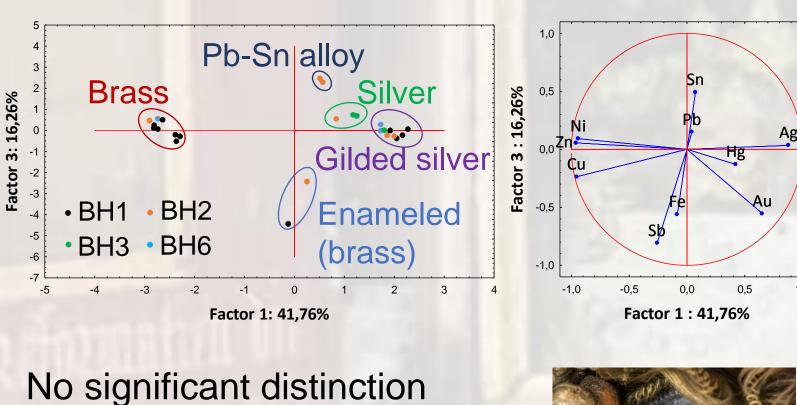
 µ-Raman SEM-EDX FTIR-ATR XRD

#### Analysis in-situ

#### Exploratory PCA

- Cu alloys (= leaded brass):
  - Back-plates
- Sequins Insignias
- 2. Lead-Tin alloys:
  - Gothic decorative elements (gilded) Insignias
- 3. Silver:
- Insignias 4. Gold leaf:
- Gilded sculptures

#### Insignias



between different panels Most silver insignias show residues of gilding

(mostly not visible)

# Factor 2: 23,87% Sequins (S) and back plates (P) Leaded brass (Cu:Zn≈80:20) NOT gilded Back plates: - BH2,BH3 → High Sn

45 -Cu/Zn -Sn (wt%) -Pb (wt%)

-Fe (wt%) -Ni (wt%)

- BH6 → Higher Cu/Zn - no technical reason (all well-preserved) Sequins: - NO differences Why so well preserved?

Factor 2: 23,87%

## Laboratory analysis: why is leaded brass so well preserved?

- NO organic protective layer (no FTIR-ATR signal)
- Protective metallic surface layer?

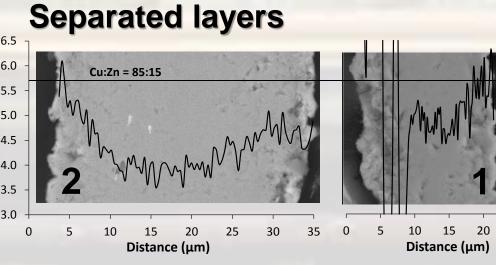
#### SEM-EDX(Surface)

	Al	Si	S	CI	K	Ca	Fe	Ni	Cu	Zn	Cu/Zn
Surface (Area)		0.1	0.1	0.1		7	0.4	0.3	84.6	14.0	6.1
Surface (Point 1)	0.3			0.1			0.3	0.2	86.3	12.7	6.8
Surface (Point 2)	0.2		0.1	0.1			0.4	0.3	85.4	13.5	6.3
Surface (Point 3)			0.1	0.1			0.2	0.2	84.6	14.6	5.8
Surface (dust)	22.0	39.6	0.5	0.9	2.1	2.6	25.8		3.9	1.0	4.1

Sequin with detached surface layer

- Well-preserved surface layer 2. Tarnished inner layer
- **Before splitting**





No Pb on the surface Higher Cu/Zn ratio on the surface than in the bulk

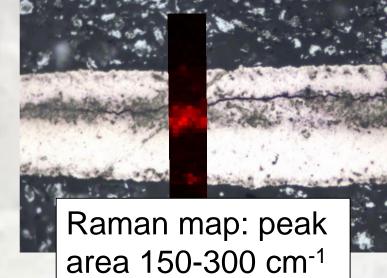
Cu:Zn ≈ 85:15 corresponds to a more golden color

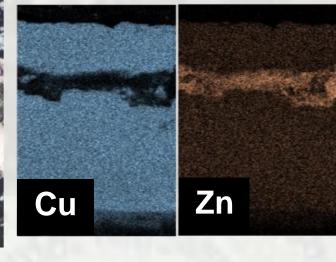
CuCl (very

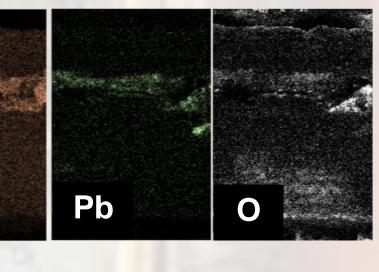
Cu:Zn = 85:15 confirmed

Discrete distribution of Pb (not soluble in the alloy) Increase of Cu/Zn towards the surface is

Degradation process (dezincification) or voluntary treatment (diluted acid) to tune the optical properties of brass?







Detached surface layer more porous and slightly higher Cu/Zn Internal corrosion layer rich in Zn, Pb and O before splitting of inner and outer layer (Metal oxides and/or chlorides - Raman). Probably consequence and not cause of the detachment.

But why no degradation products on the outer surfaces of the sample?

### CONCLUSIONS

The results of this study showed no clear and systematic difference between the metallic decorative elements in the difference between the metallic decorations were observed only in the brass back-plates decorating the background of the panels. However, these slight changes do not seem to be justified by any technological need and show no effect on the overall stability. All the lead-tin gothic elements decorating the wooden sculptures, as well as most of the brass objects showed any traces of gold. The golden appearance of brass back-plates and

sequins is in fact only related to a higher Cu/Zn ratio on the surface. It is not clear whether this dezincification is the result of a degradation processes or of a surface treatment aiming at enhancing the gold-like appearance of the metal. Contrairly to what expected, this porous Zn-depleted surface layer seems to protect the alloy from further degradation. The presence in the Gardens of large amounts of silk, potentially acting as a chemical sieve against gaseous pollutants, might also have positively influenced the overall stability of the brass elements.







